

ADDITIONS AND CORRECTIONS

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Isabel Rozas,* Ibon Alkorta, and José Elguero: Unusual Hydrogen Bonds. $H\cdots\pi$ Interactions

Page 9457. There were numerical mistakes in Table 7. Where the atomic charges (N) should appear the atomic net charges were gathered; this affects the $N(H)_{\text{complex}}$, ΔN , and $N(F)_{\text{complex}}$ columns. A revised version of Table 7 appears below.

TABLE 7: Atomic Populations and Total Atomic Energies of the HB Hydrogen Atom in the Complexes Calculated at MP2/6-311++G Level; The Total Charge Transferred (ΔQ , e) in the Formation of the Complexes is also Gathered**

	$N(H)_{\text{complex}}$	ΔN^a	$E(H)_{\text{complex}}$	ΔE^b	$N(F)_{\text{complex}}$	ΔQ^c
FH \cdots acetylene	0.290	0.002	-0.2924	0.0033	9.742	-0.032
FH \cdots ethylene	0.298	0.010	-0.2919	0.0038	9.744	-0.042
FH \cdots cyclopropene	0.296	0.008	-0.2936	0.0021	9.739	-0.035
FH \cdots cyclopropane	0.290	0.002	-0.2911	0.0046	9.741	-0.031
FH \cdots cyclobutadiene	0.300	0.012	-0.2942	0.0015	9.745	-0.045
FH \cdots tetrahydrane	0.298	0.010	-0.2925	0.0032	9.745	-0.043
FH \cdots benzene	0.288	0.000	-0.2917	0.0040	9.735	-0.024

^a $N_{\text{FH}}(H) = 0.288$. ^b $E_{\text{FH}}(H) = -0.2957$. ^c $N_{\text{FH}}(F) = 9.712$.